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STUDIES ON MONOCYCLIC β -LACTAM ANTIBIOTICS

III[†]. SYNTHESIS AND ANTIBACTERIAL ACTIVITY OF *N*-(AROMATIC HETEROCYCLIC SUBSTITUTED)AZETIDIN-2-ONES

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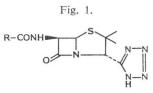
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The relationship between structure and antibacterial activity among monocyclic β -lactams having a pyridyl, pyrimidinyl, thiazolyl, imidazolyl, or a tetrazolyl group at N-1 position was investigated. *N*-(Tetrazol-5-yl)azetidin-2-ones were found to posses excellent activity.

The introduction of electron withdrawing groups at N-1 position of monocyclic β -lactam antibiotics has been extensively reported^{1~8)}. In our previous paper⁹⁾, we reported on the antibacterial activity and the β -lactamase inhibitory activity of 2-azetidinone-1-oxysulfonic acids. They had an insufficient stability against penicillinase. On the other hand, tetrazolyl penam compounds¹⁰⁾ (Fig. 1) obtained by replacing the C-3 carboxyl group of 6-aminopenicillanic acid with a tetrazolyl group, have been found to have excellent stability against β -lactamases. Thus, we attempted to synthesize monocyclic β -lactams with aromatic heterocyclic groups including a tetrazolyl group at N-1, and investigated the structure-activity relationships

regarding their antibacterial and β -lactamase inhibitory activities.

We were also interested in the effect of functional groups on the tetrazole ring and synthesized compounds 23 and 24 having a carboxymethyl group.

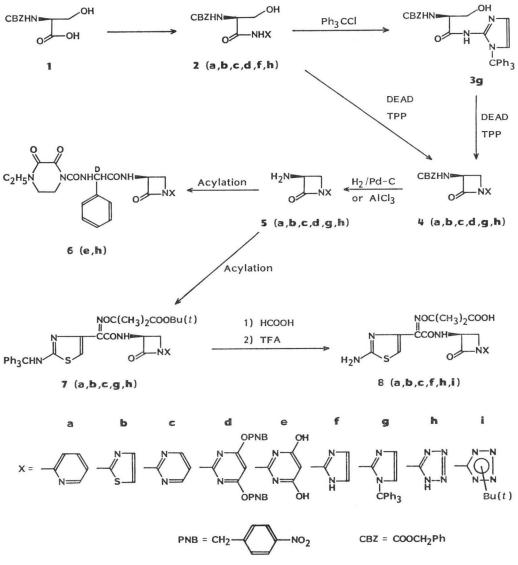


Chemistry

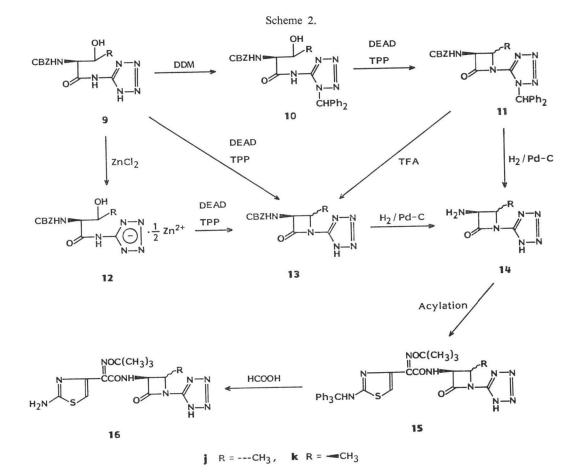
The synthesis of 3-amino-2-azetidinones (5), key intermediates for preparation of various compounds are shown in Scheme 1. The synthesis was carried out by using *N*-CBZ-L-serine (1) as a starting material. Compounds 2a, 2c, 2d and 2f were not able to be obtained by the mixed anhydride method using compound 1 and ethyl chloroformate. Instead compound 1 was silylated with trimethylchlorosilane (TMS-Cl) - Et₃N and reacted with trichloroethyl chloroformate (TCF) to obtain the chloride, which was reacted with *N*-trimethylsilyl heteroaromatic amines (TMS-HN-X (a, c, d, f)) to afford 2 in $34 \sim 46\%$ yield. MILLER *et al.*¹¹⁾ modification of the MITSUNOBU reaction¹²⁾ was very applica-

[†] Paper II. YOSHIDA, C.; T. HORI, K. MOMONOI, K. NAGUMO, J. NAKANO, T. KITANI, Y. FUKUOKA & I. SAIKAWA: Studies on monocyclic β -lactam antibiotics. II. Synthesis and antibacterial activity of 3-acylamino-2-azetidinone-1-oxysulfonic acids. J. Antibiotics 38: 1536~1549, 1985

Scheme 1.



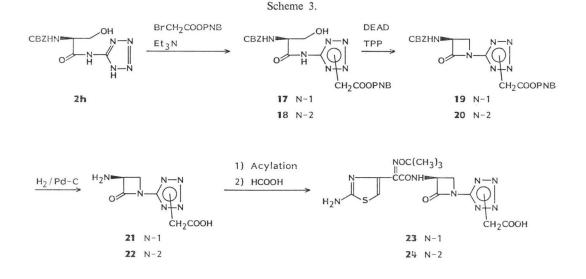
ble to the synthesis of the *N*-(aromatic heterocyclic substituted)azetidin-2-ones. However, this method could not be applied to aromatic compounds **2f** and **2h** having an active ring proton (NH group). Especially, β -lactam was not obtained at all from **2f** using diethyl azodicarboxylate (DEAD) and triphenyl phosphine (TPP). Tritylation of **2f** gave **3g** which could be cyclized to β -lactam (**4g**). **3**-Amino- β -lactams (**5**) were easily obtained from the β -lactams (**4**) by removing the CBZ group using hydrogenolysis or aluminum trichloride¹³⁾. Subsequently, **3**-acylamino- β -lactams (**6e**, **6h**) were synthesized by established acylation methods. Despite of the successful condensation of **5d** and the α -oxyiminoacetic acid, compound **7e** could not be isolated because of its instability. When compound **7h** was reacted with TFA - CH₂Cl₂ (**3**: 1) to remove the *tert*-butyl group, not only the expected compound **8h** but also **8i** having a *tert*-butyl group in the tetrazole ring were formed. Isobutene formed in the cleavage of the *tert*-butyl ester is considered to react with the tetrazole ring in presence of the



catalyst of TFA to form 8i. However, the position of the tert-butyl group was not identified.

Scheme 2 describe the synthetic route to 3-amino-4-methyl-1-(1*H*-tetrazol-5-yl)-2-azetidinones (14). Cyclization of compounds 9j and 9k obtained from *N*-CBZ-L-threonine and *N*-CBZ-L-allothreonine, respectively, in a similar manner as in Scheme 1, gave only 20% of 13. Protection of the nitrogen atom of the tetrazole ring increased the yield of the cyclization. Diphenyl diazomethane (DDM) was found to react selectively and quantitatively at the N-1 position of the tetrazole ring to afford 10, whereas the trityl group as reported by CHRISTENSEN *et al.*⁵⁾ was selectively introduced at the N-2 position. Other alkyl halides were found not to be selective regents in terms of substitution position. The protected compounds (10) were cyclized to afford β -lactams (11) in a yield of 70%. Alternatively, Zn salts (12) formed by reacting 9j and 9k with zinc chloride were directly cyclized to 13 in high yields of (63~70%). Removal of CBZ and benzhydryl groups by hydrogenolysis and TFA, respectively, gave 14, which was acylated in a similar manner as in Scheme 1. The azetidinone 26, without substituent at C-4 was synthesized in a similar way.

1-(Carboxymethyltetrazol-5-yl)azetidin-2-ones (23, 24) were prepared from compound 2h (Scheme 3). The alkylation of 2h with *p*-nitrobenzyl bromoacetate and Et_3N in dimethylformamide gave the N-1 isomer (17) and the N-2 isomer (18) in a ratio of 1 to 2. Structural confirmation of the N-1 and N-2 substituted tetrazole isomers were made according to the report of ISIDA *et al.*¹⁴⁾ in that the chemi-



cal shift of methylene group at the N-2 position was reported to appear at lower field than that at the N-1 position. The ¹H NMR spectrum of the N-1 substituted tetrazole (17) in CDCl₃ - CF₃COOD (1: 1) solution showed the resonance of the methylene protons adjacent to the tetrazole ring as a singlet at δ 5.43, while the N-2 isomer (18) appeared as a singlet at δ 5.58. The substituted compounds (17, 18) were respectively cyclized to afford β -lactams (19, 20) in 68 and 72% yield. Deprotection of 19 and 20 by hydrogenolysis of the CBZ and PNB-protecting group was followed by coupling with the α -oxyiminoacetic acid *via* the acid chloride formed with VILSMEYER-reagent. The triphenylmethyl group was removed under the mild conditions by 50% aqueous formic acid to give 23 and 24 in 41 and 47% yield, respectively.

The structure of the 3-amino- β -lactams and the 3-acylamino- β -lactams were confirmed by IR and NMR. Physical properties were listed in Tables $3 \sim 6$.

Biological Properties and Discussion

Antibacterial Activity

The minimum inhibition concentration (MIC) values of the 3-acylamino-1-(1H-tetrazol-5-yl)-2azetidinones against two staphylococci and several strains of Gram-negative bacteria were determined by the agar dilution method (Tables 1 and 2). Aztreonam¹⁵⁾ was used as a reference compound.

Table 1 shows the structure-activity relationship of N-(aromatic heterocyclic substituted)azetidin-2-ones bearing the same acyl group as that of aztreonam. Compounds $8a \sim c$ and 8f having aromatic heterocyclic groups other than tetrazole hardly showed any activity. Compounds 8h, 25, having an acidic tetrazolyl group, and aztreonam showed excellent activity against Gram-negative bacteria, but did not show activity against *Staphylococcus aureus* and *S. epidermidis*. Compounds 8h and 25 were inferior to aztreonam, against *Pseudomonas aeruginosa*. Compound 25 with a CH₃ group in the *R*configuration at C-4 showed higher activity than the unsubstituted analog (8h). The non-acidic compound 8i showed activity to medium extent.

Table 2 shows the structure-activity relationships of compounds with a CH₃ group at C-4 or a carboxymethyltetrazole at N-1. The α -(*tert*-butyloxyimino)acetyl group which has proved to show

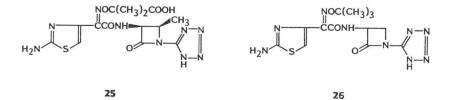


Table 1. Antibacterial (MIC μ g/ml) and cephalosporinase inhibitory activities of *N*-(aromatic heterocyclic substituted)azetidin-2-ones.

Organismsª	8a	8b	8c	8f	8h	25	8i	Aztreonam (Control)
S.e. IID 866	200	>200	>200	200	>200	>200	>200	>200
S.a. F-137*	200	>200	> 200	> 200	> 200	> 200	>200	>200
E.c. NIHJ JC-2	200	> 200	>200	> 200	0.2	0.2	12.5	0.2
E.c. TK-3*	200	> 200	> 200	> 200	0.39	≦0.1	12.5	≤ 0.1
E.c. GN 5482**	200	200	200	> 200	0.78	0.2	200	6.25
K.p. Y-50	> 200	> 200	>200	>200	≦0.1	≦0.1	6.25	≦0.1
K.p. Y-4*	200	>200	>200	>200	0.39	0.39	25	≦0.1
E.cl. IID 977	> 200	> 200	>200	> 200	≤ 0.1	≤ 0.1	200	3.13
S.m. IID 620	>200	200	> 200	> 200	6.25	0.78	3.13	≤ 0.1
S.m. W-8**	200	200	12.5	> 200	6.25	0.78	> 200	6.25
<i>P.m.</i> T-111	>200	>200	>200	> 200	≤ 0.1	≦0.1	3.13	≤ 0.1
P.v. GN 76**	200	50	200	> 200	≦0.1	≦0.1	3.13	≤ 0.1
P.a. IFO 3445	>200	>200	>200	> 200	100	12.5	>200	3.13
P.a. GN 918**	200	> 200	>200	>200	25	0.78	> 200	12.5
I ₅₀ ^b	N.T.°	160	400	N.T.	0.62	0.15	460	0.02

^a Organisms included in the Table are: S.e., Staphylococcus epidermidis; S.a., Staphylococcus aureus; E.c., Escherichia coli; K.p., Klebsiella pneumoniae; E.cl., Enterobacter cloacae; S.m., Serratia marcescens; P.m., Proteus mirabilis; P.v., Proteus vulgaris; P.a., Pseudomonas aeruginosa.

^b I₅₀ values were determined using 100 μ M cephaloridine as substrate for *E. cloacae* H-27.

° Not tested.

* Penicillinase producer. ** Cephalosporinase producer.

relatively broad spectrum of activity and the α -ureidoacetyl group of piperacillin was used in the acylamino group at C-3. Compounds with a CH₃ group at C-4 differed very little in activity against sensitive strains. Compounds 23 and 24 having carboxymethyltetrazole had no activity.

Compounds of the ureido, **6e** having a 4,6-dihydroxypyrimidinyl group at N-1 did not exhibit activity, whereas the tetrazole compound **6h** show inferior activity compared to the compound having the α -oxyiminoacetyl side chain. The change of acyl group seems to contribute more to the anti-bacterial activity in compounds with tetrazolyl groups than in the previously reported compounds with 1-oxysulfonic acid⁹⁾.

β -Lactamase Inhibitory Activity

Compounds 8h, 16j, 16k and 23~26 showed relatively strong inhibitory activity against cephalosporinase, being about 100 times as active as sulbactam¹⁷⁾ (I_{50} , 14 µg/ml not described in paper), but still inferior to aztreonam. All compounds had I_{50} values > 500 µg/ml against the penicillinase produced by *Escherichia coli*, TK-3, even though having appreciable activity against the organism. Similarly, compounds 23 and 24 exhibiting little antibacterial activity showed inhibitory activity against the cephalosporinase.

As described above, N-(tetrazol-5-yl)azetidin-2-ones derivatives possesed excellent antibacterial

Table 2. Antibacterial (MIC μ g/ml) and cephalosporinase inhibitory activities of *N*-(tetrazol-5-yl)azetidin-2-ones.

Organismsª	26	16j	16k	23	24	6e	6h	Aztreonam (Control)
S.e. IID 866	100	50	100	>200	>200	>200	6.25	>200
S.a. F-137*	>200	> 200	>200	>200	>200	>200	100	> 200
E.c. NIHJ JC-2	1.56	1.56	1.56	>200	>200	> 200	12.5	0.2
<i>E.c.</i> TK-3*	6.25	1.56	1.56	> 200	> 200	50	100	≤ 0.1
<i>E.c.</i> GN 5482**	3.13	0.78	0.39	>200	>200	200	100	6.25
K.p. Y-50	0.78	0.78	0.78	100	100	> 200	6.25	≤ 0.1
<i>K.p.</i> Y-4*	6.25	6.25	6.25	>200	>200	> 200	200	≤ 0.1
E.cl. IID 977	6.25	1.56	3.13	> 200	> 200	>200	50	3.13
S.m. IID 620	6.25	12.5	6.25	>200	> 200	200	25	≤ 0.1
S.m. W-8**	25	3.13	0.78	> 200	> 200	>200	>200	6.25
<i>P.m.</i> T-111	1.56	3.13	3.13	> 200	200	50	12.5	≦0.1
<i>P.m.</i> GN 76**	0.78	0.78	0.78	100	12.5	50	50	≦0.1
P.a. IFO 3445	> 200	200	> 200	> 200	100	>200	> 200	3.13
<i>P.a.</i> GN 918**	12.5	25	3.13	> 200	50	>200	>200	12.5
I ₅₀ ^b	0.12	0.10	0.19	0.42	0.48	N.T.°	400	0.02

^{$a \sim c$} See the footnotes in the Table 1.

activity against Gram-negative bacteria. In a forthcoming report, N-(tetrazol-5-yl)azetidin-2-ones which have substituents at C-3 and C-4 and shows activity against Gram-positive bacteria as well will be described¹⁸⁾.

Experimental

IR spectra were recorded on a Hitachi model 260-30 spectrophotometer. NMR spectra were recorded on a Hitachi R-24 (60 MHz) spectrometer using TMS as an internal standard. Melting points are uncorrected. Organic solvents were dried over anhydrous $MgSO_4$, and all concentration and evaporation of solvents were carried out under reduced pressure. Column chromatography was carried out on Wako silica gel (C-200).

In Vitro Antibacterial Activity

Minimum inhibitory concentration (MIC) was determined by the agar dilution method using heart infusion agar (Eiken). MIC (μ g/ml) was determined after incubation for 20 hours at 37°C and an inoculum size of about 10⁴ cfu.

In Vitro β -Lactamase Inhibitory Activity

I₅₀ values were determined by the method of MAK *et al.*¹⁹⁾. Experimental details for the β -lactamase inhibitory activity studies have been described previously⁹⁾.

Materials

Compounds 2h, $4h \sim 8h$ and $9 \sim 26$ were prepared as previously reported in the reference^{4,5,8,20)}. The following examples comprise a representative selection of preparative procedures and further examples may be found in the patent literature²⁰⁾.

Preparation of (2S)-2-Benzyloxycarbonylamino-3-hydroxy-N-heteroaromatic Substituted Propionamides (2)

(Method A) Preparation of 2b: To a solution of N-CBZ-L-serine (1) (10 mmol) in CH₂Cl₂ (25 ml) was added N-methylmorpholine (10 mmol) under ice-cooling. ClCOOEt (10.5 mmol) in CH₂Cl₂ (2 ml) was added dropwise to the resulting solution at $-30 \sim -20^{\circ}$ C over $10 \sim 15$ minutes, and stirred at $-20 \sim -15^{\circ}$ C for 1 hour. 2-Aminothiazole (10 mmol) in CH₂Cl₂ (7 ml) was added dropwise to the resulting mixture at $-30 \sim -20^{\circ}$ C over $15 \sim 20$ minutes and, stirred at $-15 \sim 5^{\circ}$ C for 1 hour.

Compound No.	Yield (%)	MP (°C)	¹ H NMR (Solvent) δ (J=Hz)	IR $\nu_{C=0}^{KBr}$ (cm ⁻¹)
2a	37.9	110~112	(CDCl ₃); 3.70~4.26 (2H, m), 4.50~4.90 (2H, m), 5.18 (2H, s), 6.88~7.10 (2H, m), 7.36 (5H, s), 7.68 (1H, m), 8.20~8.36 (2H, m), 10.03 (1H, br s)	1725, 1690
2b	71.9	168~170	$(DMSO-d_{e})$; 3.80 (2H, m), 4.36~4.75 (2H, m), 5.15 (2H, s), 7.29~7.64 (7H, m), 12.5 (1H, br s), 10.0 (1H, d, 8)	1720 (sh), 1680
2c	45.5	Amorphous	$(DMSO-d_{e})$; 3.45~3.84 (3H, m) 4.59 (1H, m), 5.17 (2H, s), 7.23~7.48 (6H, m), 8.25 (1H, d, 8), 8.85 (2H, d, 5), 10.55 (1H, br s)	1720, 1700
2d	34.3	126~128	(CDCl ₃); 3.56~4.07 (9H, m), 4.93~5.30 (7H, m), 5.95 (1H, s), 6.17 (1H, d, 8), 6.88 (4H, d, 9), 7.22~7.45 (9H, m), 8.90 (1H, br s)	1710, 1690
2f	43.8	182~184 (dec)	$(DMSO-d_8+D_2O)$; 3.83 (2H, m), 4.47 (1H, m), 5.15 (2H, s), 6.90 (2H, s), 7.43 (5H, s)	1690, 1670
2h	83	203~205 (dec)	$(DMSO-d_{e})$; 3.69~4.00 (2H, m), 4.44 (1H, m), 5.11 (2H, s), 7.43 (6H, m), 12.30 (1H, s)	1720, 1685
3g	28.9	Amorphous	$(CDCl_3)$; 3.24~3.63 (2H, m), 3.90 (1H, m), 5.03 (2H, s), 5.46 (1H, d, 6), 6.53 (1H, d, 2.5), 6.75 (1H, d, 2.5), 7.00~7.55 (20H, m), 7.85 (2H, br s)	1715, 1690

Table 3. Yield, mp, ¹H NMR and IR data of L-seryl heteroaromatic amides $[2(a \sim d, f, h)]$ and 3g].

The reaction mixture was evaporated to give a residue, which was dissolved in EtOAc (30 ml) and H_2O (30 ml). The organic layer was washed successively with saturated NaHCO₃ solution and brine, dried, and evaporated to give a residue, which was crystalized from 2-PrOH (20 ml) to afford **2b** in 71.9%.

(Method B) Preparation of 2a, c, d and 2f: To a solution of *N*-CBZ-L-serine (1) (10 mmol) in CH₂Cl₂ (40 ml) was added Et₃N (20.5 mmol), and added dropwise TMS-Cl (21 mmol) under icecooling, and stirred at the same temp for 1 hour. TCF (5.5 mmol) in CH₂Cl₂ (2 ml) was added dropwise to the resulting solution at $-30 \sim -25^{\circ}$ C over 10 minutes, and stirred at $-15 \sim -10^{\circ}$ C for 2 hours. Separately, heteroaromatic amines (H₂N-X (a, c, d, f)) (11 mmol) and Et₃N (11.6 mmol) in CH₂Cl₂ (10 ml) were treated dropwise with TMS-Cl (12.1 mmol) under ice-cooling, and stirred at room temp for 1 hour, and then added dropwise to the solution prepared from 1 at $-20 \sim -15^{\circ}$ C over 10 minutes, and stirred at $0 \sim 10^{\circ}$ C for 2 hours. H₂O (30 ml) was added dropwise and the mixture was extracted with CH₂Cl₂. The organic layer was treated as above to afford 2a, c, d, f in 34.3~45.5%. The physical properties of 2 are summarized in Table 3.

(2S)-2-Benzyloxycarbonylamino-3-hydroxy-N-(1-triphenylmethylimidazol-2-yl)propionamide (3g)

To a solution of **2f** (1.65 mmol) in DMF (5 ml) were added triphenylmethylchloride (1.8 mmol) and Et₃N (1.7 mmol) under ice-cooling, and stirred at $15 \sim 20^{\circ}$ C for 12 hours. H₂O (30 ml) and EtOAc (30 ml) were added into the reaction mixture. The organic layer was washed successively with H₂O and brine, dried, and evaporated to give a residue, which was purified by column chromatograpy (toluene - EtOAc, 2: 1) to afford **3g** as an amorphous powder in a yield of 28.9%.

3-Benzyloxycarbonylamino-2-azetidinones (4)

To a solution of propionamide $2a \sim d$ and 3g (1 mmol) and TPP (1.3 mmol) in THF (15 ml) was added dropwise DEAD (1.3 mmol) at $5 \sim 10^{\circ}$ C over 15 minutes and the mixture was stirred for 3 hours at room temp. The reaction mixture was evaporated and the residue dissolved in EtOAc (10 ml), washed successively with H₂O and brine, dried, and evaporated to give a residue, which was purified by column chromatography (benzene - EtOAc, 50: 1) to afford 4 in $21.5 \sim 51.3\%$ yield. The physical properties of 4 are summarized in Table 4.

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Compound No.	Yield (%)	MP (°C)	¹ H NMR (Solvent) δ (J=Hz)	$\frac{\text{IR }\nu_{C=0}^{\text{KBr}}}{(\text{cm}^{-1})}$	Anal ^a
4a	44.4	189~189.5	(DMSO- <i>d</i> ₆); 3.77 (1H, m), 4.10 (1H, m), 4.93 (1H, m), 5.17 (2H, s), 7.16~7.98 (8H, m), 8.26 (1H, d, 8), 8.48 (1H, m)	1770, 1685	$C_{16}H_{15}N_{3}O_{3}$
4b	51.3	179~180	(DMSO- d_{θ}); 3.92 (1H, m), 4.24 (1H, m), 4.98 ~ 5.27 (3H, m), 7.32 ~ 7.62 (7H, m), 8.30 (1H, d, 9)	1770, 1695	$C_{14}H_{13}N_3O_3S$
4c	28.6	148~149	(DMSO- d_{θ}); 3.81 (1H, m), 4.61 (1H, m), 4.97 (1H, m), 5.16 (2H, s), 7.24~7.50 (6H, m), 8.24 (1H, d, 8), 8.83 (2H, d, 5)	1775, 1705	$C_{15}H_{14}N_4O_3$
4d	21.5	171~172	(DMSO- d_{θ}); 3.55 ~ 3.80 (7H, m), 4.04 (1H, m), 4.93 (1H, m), 5.10 (2H, s), 5.30 (4H, s), 5.92 (1H, s), 6.97 (4H, d, 9), 7.40 ~ 7.68 (9H, m), 8.17 (1H, d, 9)	1770, 1715	$C_{\mathtt{31}}H_{\mathtt{30}}N_4O_7$
4g	50	204~208 (dec)	$(DMSO-d_{\theta})$; 2.17 (1H, m), 2.86 (1H, m), 4.09 (1H, m), 5.10 (2H, s), 6.82 (1H, d, 2), 7.09~7.72 (21H, m), 8.04 (1H, d, 8)	1780, 1725	$C_{33}H_{28}N_4O_3\\$
4h	24	153~156	(DMSO- d_6 +D ₂ O); 3.75~4.30 (2H, m), 4.95~5.31 (3H, m), 7.49 (5H, s)	1800, 1780, 1700	$C_{12}H_{12}N_6O_3$

Table 4. Yield, mp, ¹H NMR, IR and analytical data of β -lactams (4).

^a All the compounds given the formula were analyzed for C, H and N; analytical results obtained for these elements were within $\pm 0.4\%$ of calculated values.

General Preparation of 3-Amino- β -lactams 5a, c, d and 5g

 β -Lactams (4) (2 mmol) were hydrogenated in MeOH (30 ml) for 30 minutes over 10% Pd-C catalyst (500 mg) at room temp and 5 atm pressure. The catalyst was filtered and washed with MeOH. The combined filtrates were evaporated to give a residue, which was triturated with isopropyl ether to afford 5 in 30~90% yield. Crude 3-amino- β -lactams (5) were used in subsequent steps without further purification. Compound 5d was acylated immediately because of its instability.

(3S)-3-Amino-1-(thiazol-2-yl)-2-azetidinone (5b)

To a solution of **4b** (250 mg, 0.82 mmol) in CH_2Cl_2 (2.5 ml) and anisole (2.5 ml) was added dropwise AlCl₃ (550 mg, 4.12 mmol) in CH_3NO_2 at $-20 \sim -10^{\circ}C$, and stirred at $-10^{\circ}C$ for 30 minutes. The reaction mixture was dropped into a mixture of EtOAc (100 ml) and H_2O (10 ml) keeping pH at 7.0 with saturated aqueous NaHCO₃. The organic layer was washed successively with H_2O and brine, dried, and evaporated to give a residue, which was purified by column chromatography (CHCl₃ -MeOH, 30: 1) to afford **5b** as a white powder in 28.8% yield. The physical properties of **5** are summarized in Table 5.

General Procedure for the Acylation of 3-Amino- β -lactams (5)

(Method A) Synthesis of (3S)-3-[D(-)-2-(4-Ethyl-2,3-dioxo-1-piperazinecarboxamido)phenylacetamido]-1-(4,6-dihydroxypyrimidin-2-yl)-2-azetidinone (6e): To a solution of 5d (200 mg, 0.43 mmol) in DMF (4 ml) was added D(-)-(4-ethyl-2,3-dioxo-1-piperazinecarboxamido)phenylacetic acid (150 mg, 0.47 mmol) and DCC (100 mg, 0.48 mmol) under ice-cooling, and stirred for 3 hours at room temp. The precipitate was filtered off, and the filtrate was evaporated to afford a residue, which was purified by column chromatography (CHCl₃ - Me₂CO, 10: 1) to afford (3S)-3-[D-(-)-2-(4-ethyl-2,3dioxo-1-piperazinecarboxamido)phenylacetamido]-1-[4,6-di-(p-methoxybenzyloxy)pyrimidin-2-yl]-2-

Compound No.	Yield (%)	MP (°C)	¹ H NMR (Solvent) δ (J=Hz)	IR $\nu_{C=0}$ (cm ⁻¹)
5a	85.7	Amorphous	(DMSO- <i>d</i> ₀); 3.24 (2H, br s), 3.50 (1H, m), 4.05 (1H, m), 4.37 (1H, m), 7.23 (1H, m), 7.53~8.10 (2H, m), 8.48 (1H, m)	1750 (KBr)
5b	28.8	127~129	$(DMSO-d_6)$; 2.70 (2H, br s), 3.53 (1H, m), 4.08 (1H, m), 4.40 (1H, m), 7.32 (1H, d, 3), 7.40 (1H, d, 3)	1745 (KBr)
5c	87.9	90~95	$(DMSO-d_{\theta})$; 3.40~3.58 (4H, m), 4.00 (1H, m), 7.20 (1H, d, 5), 8.78 (2H, d, 5)	1780 (THF)
5g	68.5	180~181 (dec)	(DMSO- d_{θ}); 2.30 (1H, m), 2.75~3.65 (4H, m), 6.80~7.69 (17H, m)	1770 (KBr)
5h	73	179~180 (dec)	(D_2O+DCl) ; 4.23~4.68 (2H, m), 5.24 (1H, m)	1785, 1765 (KBr)

Table 5. Yield, mp, ¹H NMR and IR data of 3-amino- β -lactams (5).

azetidinone (6d) (70 mg, 21%) as a white powder. The powder was dissolved in a mixture of anisole (0.5 ml) and TFA (1.5 ml) under ice-cooling, and stirred at room temp for 30 minutes. The solvent was evaporated and the obtained residue was triturated with isopropyl ether to afford 6e as a white powder in 15% yield. Crude 6e was subjected to antibacterial activity test without further purification. MP 205~210°C (dec); IR $\nu_{C=0}^{\text{KBr}}$ 1775, 1720, 1710, 1685, 1660 cm⁻¹; NMR (DMSO- d_{θ}) δ 1.12 (3H, t, J=7.0 Hz), 3.25~4.31 (8H, m), 5.25 (1H, m), 5.55 (1H, d, J=8.0 Hz), 7.35~7.60 (8H, m), 9.56 (1H, d, J=8.0 Hz), 10.05 (1H, d, J=8.0 Hz).

(Method B) Synthesis of (3S)-3-{2-(2-Aminothiazol-4-yl)-(Z)-2-[(1-carboxy-1-methylethoxy)imino]acetamido}-1-substituted Azetidin-2-ones (8a, b, c, d and 8f): i) To a solution of 5a, b, c, g (2 mmol) in DMF (5 ml) were added successively 1-hydroxybenzotriazole hydrate (2 mmol), molecular sieves 4A (1 g), DCC (2.2 mmol) and (Z)-2-[(1-tert-butoxycarbonyl-1-methylethoxy)imino]-2-(2-triphenylmethylaminothiazol-4-yl)acetic acid (2 mmol) under ice-cooling, and stirred for 1 hour at room temp. The precipitate was filtered off, and the filtrate was evaporated. The residue was dissolved in EtOAc (20 ml), which was washed successively with H₂O and brine, dried, and evaporated to give a residue, which was purified by column chromatography to afford 7a, b, c and 7g in 50~75% yield, which were used in the following step.

ii) The compounds 7a, b, c and 7g were dissolved in THF (2 ml) and 50% aqueous HCOOH (4 ml) and kept at $40 \sim 50^{\circ}$ C for 1 hour. The solvent was evaporated to afford a residue, which was dissolved in EtOAc (20 ml) and H₂O (10 ml), and adjusted to pH 7.5 with saturated NaHCO₃ solution. The separated organic layer was washed with brine, dried, and evaporated to afford a residue, which was dissolved in CH₂Cl₂ (0.5 ml), and treated with TFA (1.5 ml) under ice-cooling. The mixture was stirred at room temp for 1.5 hours, then the solvent was evaporated and the residue triturated with isopropyl ether to afford the TFA salt of 8a, b, c, f as a white powder in $35 \sim 55\%$ yield. Crude TFA salt of 8a, b, c, f are summarised in Table 6.

To a solution of (3S)-3-{2-(2-aminothiazol-4-yl)-(Z)-2-[1-tert-butoxycarbonyl-1-methylethoxy)imino]acetamido}-1-(1*H*-tetrazol-5-yl)-2-azetidinone (1.05 g, 2.32 mmol) in CH₂Cl₂ (2.1 ml) was added TFA (6.3 ml) under ice-cooling, and stirred at room temp for 1.5 hours. The solvent was evaporated to afford a residue, which was dissolved in EtOAc (50 ml) and H₂O (50 ml), and adjusted to pH 4.0 with saturated NaHCO₃ solution. The layers were separated. The organic layer was washed with brine, dried, and evaporated to give a residue, which was purified by column chromatography (toluene -EtOAc, 1: 1) to afford **8i** as a white powder in 15% yield. The aqueous layer was adjusted to pH

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Compound No.	MP (°C, dec)	¹ H NMR (Solvent) δ (J=Hz)	$\frac{\text{IR }\nu_{C=0}^{\text{KBr}}}{(\text{cm}^{-1})}$
8a*	187~190	$(DMSO-d_0)$; 1.53 (6H, s), 3.87 (1H, m), 4.22 (1H, m), 5.26 (1H, m), 7.03 (1H, s), 7.30 (1H, m), 7.73~8.23 (7H, m), 8.50 (1H, m), 9.42 (1H, d, 8)	1755, 1705, 1660
8b*	170~174	$(DMSO-d_6)$; 1.55 (6H, s), 3.76~4.38 (2H, m), 5.28 (1H, m), 7.03 (1H, s), 7.16~7.73, (5H, m), 8.88 (2H, d, 5), 9.40 (1H, d, 7)	1775, 1740, 1660
8c*	138~142	$(DMSO-d_{\theta})$; 1.51 (6H, s), 3.97 (1H, m), 4.37 (1H, m), 5.29 (1H, m), 7.01 (1H, s), 7.40~7.65 (2H, m), 8.20 (4H, br s), 9.60 (1H, d, 8)	1775, 1720, 1660, 1630
8f*	150~155	$(DMSO-d_6+D_2O)$; 1.50 (6H, s), 3.45 (1H, m), 4.20 (1H, m), 4.97 (1H, m), 7.00 (1H, s), 7.1~7.5 (2H, m)	1790, 1715, 1665, 1630
8h	115~120	(DMSO- d_6 +D ₂ O); 1.50 (6H, s), 3.80~4.30 (2H, m), 5.30 (1H, m), 6.87 (1H, s)	1770, 1660
25*	120~140	$(DMSO-d_6+D_2O)$; 1.58 (6H, s), 1.69 (3H, d, 6.5), 4.7 (1H) ^a , 5.62 (1H, d, 5), 7.10 (1H, s)	1760, 1660
8i	183~187	$(DMSO-d_{\theta})$; 1.48 (6H, s), 1.68 (9H, s), 3.89~4.33 (2H, m), 5.28 (1H, m) 6.90 (1H, s), 7.40 (3H, br s), 9.30 (1H, d, 8)	1775, 1720, 1660
16j	145~150	$(DMSO-d_6)$; 1.30 (9H, s), 1.63 (3H, d, 7), 4.45 (1H, m), 4.96 (1H, m), 6.87 (1H, s), 7.25 (3H, br s), 9.25 (1H, d, 7)	1760, 1650, 1620
16k	130~140	$(DMSO-d_8+D_2O)$; 1.28 (9H, s), 1.46 (3H, d, 6.5), 4.65 (1H, m), 5.55 (1H, d, 5), 6.87 (1H, s)	1760, 1650, 1620
26	95~97	$(DMSO-d_{\theta})$; 1.47 (9H, s), 3.81~4.41 (2H, m), 5.41 (1H, m), 7.03 (1H, s), 9.47 (1H, d, 7) 10.1 (3H, br s)	1775, 1660

Table 6. MP, ¹H NMR and IR data of 3-acylamino-1-(aromatic heterocyclic substituted)azetidin-2-ones [8(a, b, c, f, h, i), 16(j, k), 25 and 26].

^a It was difficult to read the δ value because the signal overlapped with those of water.

* TFA salt.

2.5 with $2 \times \text{HCl}$, extracted with EtOAc, washed with brine, dried, and evaporated to give a residue, which was triturated with Et_2O to afford **8h** as a white powder in 60% yield. The physical properties are summarized in Table 6.

(2S,3S)-2-Benzyloxycarbonylamino-3-hydroxy-N-(1-diphenylmethyl-1H-tetrazol-5-yl)butyramide (10k)

To a solution of **9** (10 g, 31.2 mmol) in MeOH (50 ml) and acetone (50 ml) was added diphenyl diazomethane (11 g, 57 mmol) at room temp, and stirred for 24 hours at the same temp. The reaction mixture was evaporated and the residue was dissolved in EtOAc (200 ml) and H₂O (100 ml), which was adjusted to pH 7.0 with saturated NaHCO₃ solution. The organic layer was washed successively with H₂O and brine, dried, and evaporated to give a residue, which was purified by column chromatography (benzene - EtOAc, 50: 1) to afford **10k** as an amorphous powder in 80% yield. IR $\nu_{C=0}^{\text{KBr}}$ 1720, 1700 cm⁻¹; NMR (DMSO- d_6 + D₂O) δ 1.31 (3H, d, J=6.0 Hz), 3.39 (1H, m), 4.30 (1H, m), 5.14 (2H, s), 7.04 (1H, s), 7.37 (15H, s).

(3S.4R)-3-Benzyloxvcarbonylamino-1-(1-diphenylmethyl-1*H*-tetrazol-5-vl)-4-methyl-2-azetidinone (11k)

To a solution of 10k (17.2 g, 35.4 mmol) and TPP (12 g, 46 mmol) in THF (400 ml) was added dropwise DEAD (7.2 ml, 46 mmol) at $5 \sim 10^{\circ}$ C over 15 minutes, and stirred for 3 hours at room temp. The reaction mixture was evaporated and the residue dissolved in EtOAc (100 ml) and H₂O (50 ml). The organic layer was washed successively with H₂O and brine, dried, and evaporated to give a residue, which was triturated with benzene (80 ml). The solid was filtered off, and the filtrate was evaporated. The residue was purified by column chromatography (benzene - EtOAc, 50: 1) to afford **11k** as colorless crystals in 70% yield. MP 158~160°C; IR $\nu_{C=0}^{\text{KBr}}$ 1780, 1720 cm⁻¹; NMR (CDCl₃) δ 1.31 (3H, d, J = 6.0 Hz), 4.54 (1H, m), 5.16~5.25 (3H, m), 5.71 (1H, d, J = 9.0 Hz), 7.38 (5H, s), 7.42 (10H, s), 7.77 (1H, s).

(3S,4R)-Benzyloxycarbonylamino-4-methyl-1-(1H-tetrazol-5-yl)-2-azetidinone (13k)

Method A: To a solution of **11k** (3 g, 6.4 mmol) in anisole (15 ml) was added TFA (45 ml) at room temp, and stirred for 2 hours. The solvent was evaporated to afford a residue, which was dissolved in Et₂O (20 ml) and ice-water (2 ml) and adjusted to pH 7.5 with saturated NaHCO₃ solution. The aqueous layer was adjusted to pH 2.0 with 2 N HCl, and extracted with EtOAc. The organic layer was washed successively with H₂O and brine, dried, and evaporated to give a residue, which was triturated with a mixture of EtOAc (5 ml) and Et₂O (10 ml) to afford **13k** as colorless crystals in 80% yield. MP 147~149°C; IR $\nu_{\text{C}=0}^{\text{T}=6}$ 1770, 1720 cm⁻¹; NMR (CDCl₃+D₂O) δ 1.52 (3H, d, J=6.0 Hz), 4.67 (1H, t, J=6.0 Hz), 5.19 (2H, s), 5.37 (1H, d, J=6.0 Hz), 7.42 (5H, s).

Method B: Cyclization of 9k was carried out in a similar manner as for 11k. Yield 20%.

Method C: To a solution of 12k (2.8 g, 7.1 mmol) in THF (70 ml) and acetone (70 ml) was added anhydrous MgSO₄ (5 g), and stirred for 30 minutes. The solid was filtered off, and the filtrate was evaporated to dryness. The residue was dissolved in THF (120 ml), TPP (2.69 g, 10.3 mmol) added, followed by DEAD (1.62 ml, 10.3 mmol) in THF (10 ml) dropwise at $20 \sim 25^{\circ}$ C over 30 minutes. The reaction mixture was stirred for 12 hours at room temp, and evaporated to give a residue, which was triturated with a mixture of EtOAc (20 ml) and Et₂O (20 ml). The separating crystals were collected by filtration and suspended in EtOAc (30 ml) and H₂O (30 ml), which was adjusted to pH 1.5 with $6 \times$ HCl. The organic layer was separated and added to H₂O (30 ml), and adjusted to pH 7.0 with saturated NaHCO₃ solution. The aqueous layer was separated, added to EtOAc (30 ml), and adjusted to pH 2.0 with $2 \times$ HCl. The organic layer was washed successively with H₂O (10 ml) and brine (10 ml), dried, and evaporated to give a residue, which was triturated with Et₂O (20 ml) to afford **13k** in 63% yield.

(3S,4R)-3-Amino-4-methyl-1-(1H-tetrazol-5-yl)-2-azetidinone (14k)

Method A: The compound 13k (2 g, 6.6 mmol) was hydrogenated in MeOH (50 ml) for 1.5 hours over 5% Pd-C (0.5 g) at atmospheric pressure and room temp. The catalyst was filtered and washed with aqueous NaHCO₃ (0.83 g, 9.9 mmol) solution (20 ml). The combined filtrates were treated with $6 \times HCl$ (2 ml) and evaporated to give a residue, which was dissolved in hot EtOH (30 ml). The solid was filtered off and the filtrate cooled to room temp to afford the HCl salt of 14k (1 g) as colorless crystals in 74.1% yield. To a suspention of the HCl salt (1 g, 4.9 mmol) in EtOH (40 ml) pyridine (0.6 ml, 7.4 mmol) was added, and stirred for 5 hours at room temp. The precipitate was collected by filtration and washed with EtOH to afford 14k as a white powder in 83% yield. MP 192°C (dec); IR $\nu_{C=0}^{KBr}$ 1810, 1775 cm⁻¹; NMR (D₂O+DCl) δ 1.72 (3H, d, J=7.0 Hz), 4.64~ 5.09 (2H, m).

Method B: The compound 11k (1 g, 2.1 mmol) was hydrogenated in MeOH (20 ml) for 5 hours over 10% Pd-C (0.5 g) at room temp and 5 atm pressure. The title compound was obtained in a similar manner as above in 30% yield.

Zinc Salt of (2S, 3S)-2-Benzyloxycarbonylamino-3-hydroxy-N-(1H-tetrazol-5-yl)butyramide· $\frac{1}{2}H_2O(12k)$

To a suspension of 9k (3 g, 9.4 mmol) in H₂O (60 ml) was added NaHCO₃ (0.83 g, 9.9 mmol) to obtain a clear solution. ZnCl₂ (0.77 g, 5.65 mmol) was added, and stirred for 30 minutes at room temp. The crystals formed were collected by filtration and washed with H₂O (10 ml) to afford 12k as white crystals in 97.3% yield. MP > 240°C; IR $\nu_{C=0}^{KBr}$ 1700, 1670 cm⁻¹.

(3S,4R)-4-Methyl-1-(1H-tetrazol-5-yl)-3-[2-(2-triphenylmethylaminothiazol-4-yl)-(Z)-2-tert-butoxy-iminoacetamido]-2-azetidinone (15k)

To a solution of 14k (0.34 g, 2 mmol) in DMF (3 ml) was added Et_3N (0.28 ml, 2 mmol) under ice-cooling. Molecular sieves 4A (0.5 g), 1*H*-hydroxybenzotriazole hydrate (0.27 g, 2 mmol), (*Z*)-2-*tert*-butoxyimino-2-(2-triphenylmethylaminothiazol-4-yl)acetic acid (1 g, 2 mmol) and DCC (0.46 g,

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2.2 mmol) were added successively to the solution at the same temp, and stirred for 2 hours at room temp. The precipitate was filtered off and the filtrate evaporated. The residue was dissolved in H₂O (15 ml) and EtOAc (15 ml) and adjusted to pH 2.0 with 2 N HCl. The organic layer was washed with brine, dried, and evaporated to give a residue, which was purified by column chromatography (CHCl₃ - MeOH, 50: 1) to afford **15k** as a white powder in 49% yield. MP 155~175°C (dec); IR $\nu_{C=0}^{KBr}$ 1765, 1660 cm⁻¹; NMR (DMSO-*d*₆) δ 1.22 (9H, s), 1.38 (3H, d, *J*=6.0 Hz), 4.48 (1H, m), 5.28~5.55 (2H, m), 6.68 (1H, s), 7.31 (15H, s), 9.26 (1H, d, *J*=9.0 Hz).

$\frac{(3S, 4R)-3-[2-(2-Aminothiazol-4-yl)-(Z)-2-tert-butoxyiminoacetamido]-4-methyl-1-(1H-tetrazol-5-yl)-2-azetidinone (16k)$

The compound **15k** (0.38 g, 0.6 mmol) was dissolved in a mixture of THF (2 ml) and 50% aqueous HCOOH (4 ml) and kept at 40~50°C for 1 hour. The solvent was evaporated to afford a residue, which was dissolved in EtOAc (10 ml) and H₂O (10 ml), and adjusted to pH 7.5 with saturated NaHCO₃ solution. The aqueous layer was adjusted to pH 2.5 with 2 N HCl, saturated with NaCl, and extrated with EtOAc (10 ml) twice. The combined extracts were washed with brine (10 ml), and evaporated to give a residue, which was triturated with Et₂O to afford **16**k as a white powder in 55% yield. MP 130~140°C (dec); IR $\nu_{\text{CBD}}^{\text{KBD}}$ 1760, 1650, 1620 cm⁻¹; NMR (DMSO-d₆+D₂O) δ 1.29 (9H, s), 1.45 (3H, d, *J*=6.0 Hz), 4.65 (1H, m), 5.55 (1H, d, *J*=5.0 Hz), 6.87 (1H, s).

(2S)-2-Benzyloxycarbonylamino-3-hydroxy-N-[1-(4-nitrobenzyloxycarboxymethyl)-1H-tetrazol-5yl]propionamide (17) and N-2 Isomer (18)

To a solution of **2h** (3 g, 0.98 mmol) in DMF (15 ml) and Et₃N (1.35 ml, 1 mmol) was added *p*-nitrobenzyl bromoacetate (2.74 g, 1 mmol) at room temp, and stirred for 4 hours. EtOAc (300 ml) and H₂O (200 ml) was added and the organic layer was separated, washed successively with H₂O and brine, dried, and evaporated to give a residue, which was purified by column chromatography (CHCl₃ - acetone, 10: 1) to afford **17** (0.95 g, 19.4%) and **18** (2.05 g, 41.9%).

17: MP 117~119°C (EtOAc - Et₂O); Rf 0.09 [Merck Silica gel 60 F_{254} , (CHCl₃ - acetone, 2:1)]; IR $\nu_{C=0}^{\text{KBr}}$ 1740, 1700 cm⁻¹; NMR (CDCl₃ - CF₃COOD, 1:1) δ 4.0~4.3 (2H, m), 4.55~4.85 (1H, m), 5.20 (2H, s), 5.36 (2H, s), 5.43 (2H, s), 7.30 (5H, s), 7.52 (2H, d, J=9.0 Hz), 8.21 (2H, d, J=9.0 Hz).

 $\begin{array}{rl} \mbox{Anal Calcd for $C_{21}H_{21}N_7O_8$:} & C $50.50, H $4.24, N $19.63.$ \\ \mbox{Found:} & C $50.61, H $4.22, N $19.35.$ \\ \end{array}$

18: MP 164 ~ 165°C (EtOAc - Et₂O); Rf 0.29 [Merck Silica gel 60 F_{254} , (CHCl₃ - acetone, 2: 1)]; IR $\nu_{C=0}^{KBr}$ 1750, 1680 cm⁻¹; NMR (CDCl₃ - CF₃COOD, 1: 1) δ 4.1~4.33 (2H, m), 4.6~4.87 (1H, m), 5.21 (2H, s), 5.39 (2H, s), 5.58 (2H, s), 7.28 (5H, s), 7.55 (2H, d, J=9.0 Hz), 8.23 (2H, d, J=9.0 Hz).

Anal Calcd for $C_{21}H_{21}N_7O_8$: C 50.50, H 4.24, N 19.63.

Found: C 50.18, H 4.31, N 19.88.

(3S)-3-Benzyloxycarbonylamino-1-[1-(4-nitrobenzyloxycarboxymethyl)-1*H*-tetrazol-5-yl]-2-azetidinone (19) and N-2 Isomer (20)

Cyclization of 19 and 20 were carried out in a similar manner as for 11k.

19: Yield 68%; mp 138~140°C (EtOAc - Et₂O); IR $\nu_{C=0}^{CH_2Cl_2}$ 1785, 1765, 1730 cm⁻¹; NMR (DMSOd₆) δ 3.76~4.20 (2H, m), 4.90~5.20 (3H, m), 5.40 (2H, s), 5.94 (2H, s), 7.37 (5H, s), 7.69 (2H, d, J= 10.0 Hz), 8.27 (2H, d, J=10.0 Hz), 8.65 (1H, d, J=8.0 Hz).

Anal Calcd for $C_{21}H_{19}N_7O_7$: C 52.39, H 3.98, N 20.37.

Found: C 52.31, H 3.77, N 20.51.

20: Yield 72%; mp 122~126°C (EtOAc - Et₂O); IR $\nu_{C=0}^{CH_2Cl_2}$ 1790, 1760, 1705 cm⁻¹; NMR (DMSO- d_0) δ 3.95~4.20 (2H, m), 5.10~5.26 (3H, m), 5.40 (2H, s), 5.92 (2H, s), 7.35 (5H, s), 7.72 (2H, d, J=10.0 Hz), 8.25 (2H, d, J=10.0 Hz), 8.60 (1H, d, J=8.0 Hz).

Anal Calcd for $C_{21}H_{19}N_7O_7$: C 52.39, H 3.98, N 20.37. Found: C 52.58, H 3.91, N 20.61.

(3S)-3-Amino-1-(1-carboxymethyl-1H-tetrazol-5-yl)-2-azetidinone (21) and N-2 Isomer (22)

Compound 19 (50 mg, 0.1 mmol) was hydrogenated in MeOH (5 ml) for 30 minutes over 10%

Pd-C (10 mg) at room temp and 5 atm pressure. The catalyst was filtered and washed with MeOH. The combined filtrates were evaporated to give a residue, which was triturated with EtOH to afford **21** as a white powder in 77% yield. MP 172~174°C (dec); IR $\nu_{C=0}^{KBr}$ 1780, 1610 cm⁻¹; NMR (D₂O) δ 4.20 (2H, m), 5.10 (1H, m), 5.40 (2H, s).

Compound 22 was synthesized in a similar manner as above. Yield 95%; mp 195°C (dec); IR $\nu_{C=0}^{KBr}$ 1785, 1635 cm⁻¹; NMR (D₂O) δ 4.30 (2H, m), 5.07 (1H, m), 5.36 (2H, s).

(3S)-3-[2-(2-Aminothiazol-4-yl)-(Z)-2-(*tert*-butoxyimino)acetamido]-1-(1-carboxymethyl-1*H*-tetrazol-5-yl)-2-azetidinone (23) and N-2 Isomer (24)

To a solution of (Z)-2-tert-butoxyimino-2-(2-triphenylmethylaminothiazol-4-yl)acetic acid (0.97 g, 2 mmol) in DMF (5 ml) was added dropwise a solution of POCl₃ (0.19 ml, 2 mmol) in CH₂Cl₂ (1 ml) at $-15 \sim -10^{\circ}$ C over 5 minutes, and stirred for 30 minutes at that temp. Separately, a mixture of **21** (0.42 g, 2 mmol) and Et₃N (0.28 ml, 2 mmol) in DMF (5 ml) was stirred for 30 minutes at the same temp to give a solution, which was cooled to -20° C and dropped into the above reaction mixture. After stirring for 1 hour at room temp, EtOAc (20 ml) and H₂O (20 ml) was added. The organic layer was washed successively with H₂O and brine, dried, and evaporated to give a residue, which was purified by column chromatography (CHCl₃ - MeOH, 50: 1) to afford a white powder (0.75 g). These materials were dissolved in THF (4 ml) and 50% aqueous HCOOH (8 ml) and kept at 40~50°C for 1 hour. The solvent was evaporated to afford a residue, which was triturated with Et₂O to afford **23** as an amorphous powder in 41% yield. IR $\nu_{C=0}^{RB}$ 1775, 1650, 1620 cm⁻¹; NMR (DMSO-d₈) δ 1.35 (9H, s), 4.10 (2H, m), 5.05~6.05 (5H, m), 6.65 (1H, s), 7.40 (1H, br s), 9.20 (1H, br s).

Compound **24** was synthesized in a similar manner as above. Yield 47%; mp 197~201°C (dec); IR $\nu_{C=0}^{\text{KBr}}$ 1775, 1650, 1620 cm⁻¹; NMR (DMSO- d_{θ}) δ 1.32 (9H, s), 4.02 (2H, m), 4.82~5.92 (5H, m), 6.83 (1H, s), 7.32 (1H, br s), 9.12 (1H, br s).

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